electronic reprint

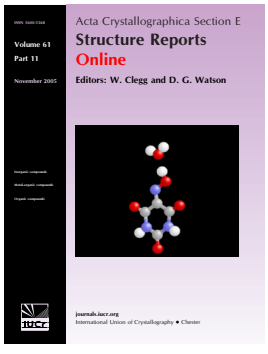
Acta Crystallographica Section E
Structure Reports Online
ISSN 1600-5368
Editors: **W. T. A. Harrison, J. Simpson and M. Weil**

N-(4,6-Dimethylpyrimidin-2-yl)-4-(oxolan-2-ylamino)benzenesulfonamide

Hadi D. Arman, Trupta Kaulgud, Edward R. T. Tiekink and David J. Young

Acta Cryst. (2009). E**65**, o2851

This open-access article is distributed under the terms of the Creative Commons Attribution Licence http://creativecommons.org/licenses/by/2.0/uk/legalcode, which permits unrestricted use, distribution, and reproduction in any medium, provided the original authors and source are cited.



Acta Crystallographica Section E: Structure Reports Online is the IUCr's highly popular open-access structural journal. It provides a simple and easily accessible publication mechanism for the growing number of inorganic, metal-organic and organic crystal structure determinations. The electronic submission, validation, refereeing and publication facilities of the journal ensure very rapid and high-quality publication, whilst key indicators and validation reports provide measures of structural reliability. In 2007, the journal published over 5000 structures. The average publication time is less than one month.

Crystallography Journals Online is available from **journals.iucr.org**

Acta Crystallographica Section E
Structure Reports Online
ISSN 1600-5368

N-(4,6-Dimethylpyrimidin-2-yl)-4-(oxolan-2-ylamino)benzenesulfonamide

Hadi D. Arman,[a] Trupta Kaulgud,[a] Edward R. T. Tiekink[b]* and David J. Young[c]

[a]Department of Chemistry, The University of Texas at San Antonio, One UTSA Circle, San Antonio, Texas 78249-0698, USA, [b]Department of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia, and [c]School of Biomolecular and Physical Sciences, Nathan, Griffith University, Queensland 4111, Australia
Correspondence e-mail: edward.tiekink@gmail.com

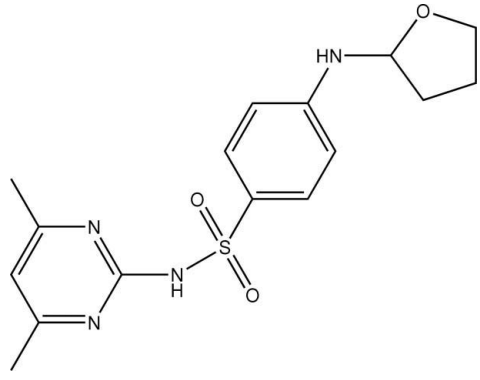
Received 20 October 2009; accepted 21 October 2009

Key indicators: single-crystal X-ray study; T = 98 K; mean σ(C–C) = 0.004 Å; R factor = 0.057; wR factor = 0.137; data-to-parameter ratio = 16.7.

The title compound, $C_{16}H_{20}N_4O_3S$, adopts an L-shaped conformation, as seen by the dihedral angle of 76.93 (7)° formed between the two aromatic rings. The most notable feature of the crystal packing is the formation of N—H···O and N—H···N hydrogen bonds that lead to supramolecular chains orientated along the b axis.

Related literature

For background to the co-crystallization of active pharmaceutical agents, see: Shan & Zaworotko (2008). For background to sulfa drugs, see: Caira (2007); Nishimori *et al.* (2009). For the synthesis, see: Fructos *et al.* (2006); Kemnitz *et al.* (1998). For related studies on co-crystal formation, see: Broker & Tiekink (2008); Broker *et al.* (2008).



Experimental

Crystal data

$C_{16}H_{20}N_4O_3S$ — M_r = 348.42
Monoclinic, $P2_1/c$
a = 10.291 (5) Å
b = 9.592 (4) Å
c = 17.196 (8) Å
β = 106.445 (10)°
V = 1628.0 (13) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 0.22 mm^{-1}
T = 98 K
0.35 × 0.21 × 0.11 mm

Data collection

Rigaku Saturn724 diffractometer
Absorption correction: multi-scan (*ABSCOR*; Higashi, 1995)
T_{min} = 0.761, T_{max} = 1.000
11164 measured reflections
3749 independent reflections
3341 reflections with $I > 2\sigma(I)$
R_{int} = 0.046

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.057
$wR(F^2)$ = 0.137
S = 1.10
3749 reflections
225 parameters
2 restraints
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.60 e Å^{-3}
$\Delta\rho_{min}$ = −0.39 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
N3—H3n···O3^{i}	0.88	1.98	2.854 (3)	174
N4—H4n···N2^{ii}	0.88	2.22	3.086 (3)	167

Symmetry codes: (i) $x, y-1, z$; (ii) $x, y+1, z$.

Data collection: *CrystalClear* (Rigaku/MSC, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *DIAMOND* (Brandenburg, 2006); software used to prepare material for publication: *SHELXL97*.

The Queensland Department of Employment, Economic Development and Innovation is thanked for an International Fellowship to DJY.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB5159).

References

Brandenburg, K. (2006). *DIAMOND*. Crystal Impact GbR, Bonn, Germany.
Broker, G. A., Bettens, R. P. A. & Tiekink, E. R. T. (2008). *CrystEngComm*, **10**, 879–887.
Broker, G. A. & Tiekink, E. R. T. (2008). *CrystEngComm*, **9**, 1096–1109.
Caira, M. R. (2007). *Mol. Pharm.* **4**, 310–316.
Fructos, M. R., Trofimenko, S., Mar Díaz-Requejo, M. & Pérez, P. J. (2006). *J. Am. Chem. Soc.* **128**, 11784–11791.
Higashi, T. (1995). *ABSCOR*. Rigaku Corporation, Tokyo, Japan.
Kemnitz, C. R., Karney, W. L. & Borden, W. T. (1998). *J. Am. Chem. Soc.* **120**, 3499–3503.
Nishimori, I., Minakuchi, T., Vullo, D., Scozzafava, A., Innocenti, A. & Supuran, C. T. (2009). *J. Med. Chem.* **52**, 3116–3120.
Rigaku/MSC (2005). *CrystalClear*. Rigaku/MSC Inc. The Woodlands, Texas, USA.
Shan, N. & Zaworotko, M. J. (2008). *Drug Discovery Today*, **13**, 440–446.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.

electronic reprint

supplementary materials

electronic reprint

Acta Cryst. (2009). E**65**, o2851 [**doi:10.1107/S1600536809043347**]

N-(4,6-Dimethylpyrimidin-2-yl)-4-(oxolan-2-ylamino)benzenesulfonamide

H. D. Arman, T. Kaulgud, E. R. T. Tiekink and D. J. Young

Comment

The co-crystallization of active pharmaceutical ingredients is an active area of contemporary crystal engineering (Shan & Zaworotko, 2008). Sulfonamide drugs, *e.g.* sulfadimidine and sulfameter, attract significant interest in this regard, especially owing to their propensity to form polymorphs (Caira, 2007). They are also receiving renewed attention as selective inhibitors of carbonic anhydrase isoforms (*e.g.* Nishimori *et al.*, 2009). As a continuation of studies into the phenomenon of co-crystallization (Broker & Tiekink, 2008; Broker *et al.*, 2008), the co-crystallization of *N*'-(4,6-dimethyl-2-pyrimidinyl)sulfanilamide (sulfadimidine) and 1,4-$C_6H_4I_2$ in THF was investigated. Colourless crystals of the title compound (I) were obtained unexpectedly; we are not aware of any precedence for this reaction. The insertion of nitrenes into the α C—H bond of cyclic ethers is known (Fructos *et al.*, 2006) and it is suggested that adventitious I_2 in 1,4-$C_6H_4I_2$ reacts with the aryl amine to give a nitrene stabilized by the *para*-sulfonamide group (Kemnitz *et al.*, 1998).

The molecule of (I), Fig. 1, is bent at the S atom, N3—S1—C7 = 107.85 (10)°, and adopts an overall `L'-conformation; the dihedral angle between the two six-membered rings is 76.93 (7)°. The five membered ring adopts an envelope configuration at the C16 atom. The crystal packing is dominated by N—H···O and N—H···N hydrogen bonding interactions, Table 1, that co-operate to form a supramolecular chain along the *b* axis, Fig. 2.

Experimental

Colourless crystals of (I) were isolated from the attempted co-crystallization of *N*'-(4,6-dimethyl-2-pyrimidinyl)-sulfanilamide and 1,4-di-iodobenzene in THF.

Refinement

Carbon-bound H-atoms were placed in calculated positions (C–H 0.95–1.00 Å) and were included in the refinement in the riding model approximation with U_{iso}(H) set to 1.2–1.5U_{eq}(C). The nitrogen-bound H-atoms were located in a difference Fourier map and were refined with a N–H 0.880±0.001 Å restraint, and with U_{iso}(H) = 1.2U_{eq}(N).

Figures

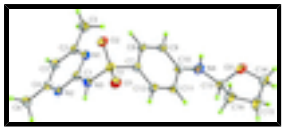


Fig. 1. Molecular structure of (I) showing atom-labelling scheme and displacement ellipsoids at the 50% probability level.

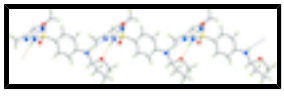


Fig. 2. Supramolecular chain formation along the *b* axis in (I) mediated by N—H···N (orange dashed lines) and N—H···N (blue dashed lines) hydrogen bonding.

electronic reprint

N-(4,6-Dimethylpyrimidin-2-yl)-4-(oxolan-2-ylamino)benzenesulfonamide

Crystal data

$C_{16}H_{20}N_4O_3S$	F_{000} = 736
M_r = 348.42	D_x = 1.422 Mg m^{-3}
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation, λ = 0.71073 Å
Hall symbol: -P 2ybc	Cell parameters from 6601 reflections
a = 10.291 (5) Å	θ = 2.5–40.2º
b = 9.592 (4) Å	μ = 0.22 mm^{-1}
c = 17.196 (8) Å	T = 98 K
β = 106.445 (10)º	Block, colourless
V = 1628.0 (13) Å^3	0.35 × 0.21 × 0.11 mm
Z = 4	

Data collection

Saturn724 diffractometer	3749 independent reflections
Radiation source: sealed tube	3341 reflections with $I > 2\sigma(I)$
Monochromator: graphite	R_{int} = 0.046
Detector resolution: 28.5714 pixels mm^{-1}	θ_{max} = 27.5º
T = 98 K	θ_{min} = 2.1º
ω scans	$h = -12 \rightarrow 13$
Absorption correction: Multi-scan (ABSCOR; Higashi, 1995)	$k = -12 \rightarrow 11$
T_{min} = 0.761, T_{max} = 1.000	$l = -22 \rightarrow 17$
11164 measured reflections	

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)]$ = 0.057	H-atom parameters constrained
$wR(F^2)$ = 0.137	$w = 1/[\sigma^2(F_o^2) + (0.0464P)^2 + 1.4631P]$ where $P = (F_o^2 + 2F_c^2)/3$
S = 1.10	$(\Delta/\sigma)_{max} < 0.001$
3749 reflections	$\Delta\rho_{max}$ = 0.60 e Å^{-3}
225 parameters	$\Delta\rho_{min}$ = −0.39 e Å^{-3}
2 restraints	Extinction correction: none
Primary atom site location: structure-invariant direct methods	

electronic reprint

Special details

Geometry. All s.u.'s (except the s.u. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell s.u.'s are taken into account individually in the estimation of s.u.'s in distances, angles and torsion angles; correlations between s.u.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell s.u.'s is used for estimating s.u.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2, conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2. The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
S1	0.31473 (6)	0.75837 (6)	0.11287 (3)	0.02795 (16)
O1	0.23391 (18)	0.71472 (18)	0.03422 (9)	0.0357 (4)
O2	0.45898 (17)	0.76422 (18)	0.12920 (9)	0.0339 (4)
O3	0.03413 (17)	1.4900 (2)	0.10040 (10)	0.0406 (4)
N1	0.40297 (19)	0.74868 (19)	0.29276 (10)	0.0271 (4)
N2	0.27644 (18)	0.53811 (19)	0.29370 (10)	0.0265 (4)
N3	0.2755 (2)	0.6434 (2)	0.17341 (10)	0.0283 (4)
H3N	0.2048	0.5906	0.1512	0.034*
N4	0.1102 (2)	1.3038 (2)	0.18799 (14)	0.0421 (5)
H4N	0.1658	1.3594	0.2227	0.051*
C1	0.3215 (2)	0.6443 (2)	0.25786 (12)	0.0258 (4)
C2	0.4447 (2)	0.7458 (2)	0.37479 (12)	0.0283 (5)
C3	0.4038 (2)	0.6408 (2)	0.41786 (12)	0.0298 (5)
H3	0.4336	0.6395	0.4754	0.036*
C4	0.3183 (2)	0.5378 (2)	0.37529 (12)	0.0283 (4)
C5	0.5376 (3)	0.8614 (3)	0.41506 (14)	0.0378 (5)
H5A	0.6280	0.8455	0.4085	0.057*
H5B	0.5435	0.8637	0.4729	0.057*
H5C	0.5021	0.9506	0.3901	0.057*
C6	0.2706 (3)	0.4199 (3)	0.41651 (14)	0.0348 (5)
H6A	0.1741	0.4032	0.3904	0.052*
H6B	0.2840	0.4432	0.4737	0.052*
H6C	0.3223	0.3357	0.4125	0.052*
C7	0.2564 (2)	0.9212 (2)	0.13414 (12)	0.0279 (4)
C8	0.3417 (2)	1.0138 (2)	0.18760 (13)	0.0291 (5)
H8	0.4334	0.9891	0.2129	0.035*
C9	0.2929 (2)	1.1411 (2)	0.20371 (13)	0.0307 (5)
H9	0.3513	1.2038	0.2402	0.037*
C10	0.1574 (2)	1.1793 (2)	0.16669 (13)	0.0318 (5)
C11	0.0733 (2)	1.0854 (2)	0.11240 (14)	0.0344 (5)
H11	−0.0181	1.1100	0.0862	0.041*
C12	0.1224 (2)	0.9580 (2)	0.09687 (13)	0.0322 (5)
H12	0.0645	0.8949	0.0605	0.039*

electronic reprint

C13	−0.0074 (3)	1.3735 (3)	0.14064 (16)	0.0384 (6)
H13	−0.0624	1.3077	0.0991	0.046*
C14	−0.0727 (3)	1.5920 (3)	0.0855 (2)	0.0582 (8)
H14A	−0.1141	1.6035	0.0265	0.070*
H14B	−0.0360	1.6832	0.1085	0.070*
C15	−0.1763 (3)	1.5425 (4)	0.1246 (2)	0.0573 (8)
H15A	−0.2534	1.4971	0.0848	0.069*
H15B	−0.2105	1.6199	0.1513	0.069*
C16	−0.0957 (3)	1.4381 (4)	0.1863 (2)	0.0618 (9)
H16A	−0.0415	1.4851	0.2362	0.074*
H16B	−0.1557	1.3681	0.2007	0.074*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0279 (3)	0.0329 (3)	0.0212 (2)	0.0002 (2)	0.0038 (2)	0.00097 (19)
O1	0.0393 (9)	0.0426 (10)	0.0218 (7)	−0.0014 (8)	0.0032 (7)	−0.0022 (6)
O2	0.0289 (8)	0.0431 (9)	0.0297 (8)	0.0015 (7)	0.0085 (7)	0.0009 (7)
O3	0.0274 (8)	0.0554 (11)	0.0384 (9)	0.0039 (8)	0.0083 (7)	0.0090 (8)
N1	0.0287 (9)	0.0260 (9)	0.0242 (8)	0.0000 (7)	0.0037 (7)	−0.0023 (7)
N2	0.0261 (9)	0.0278 (9)	0.0245 (8)	0.0005 (7)	0.0053 (7)	0.0001 (7)
N3	0.0307 (10)	0.0284 (9)	0.0221 (8)	−0.0028 (8)	0.0013 (7)	−0.0012 (7)
N4	0.0332 (11)	0.0337 (11)	0.0476 (12)	0.0010 (9)	−0.0079 (9)	−0.0087 (9)
C1	0.0250 (10)	0.0260 (10)	0.0246 (9)	0.0039 (8)	0.0042 (8)	−0.0011 (8)
C2	0.0297 (11)	0.0273 (10)	0.0251 (10)	0.0030 (9)	0.0031 (8)	−0.0039 (8)
C3	0.0329 (11)	0.0349 (12)	0.0198 (9)	0.0012 (9)	0.0044 (8)	−0.0017 (8)
C4	0.0277 (11)	0.0310 (11)	0.0265 (10)	0.0024 (9)	0.0080 (8)	0.0005 (8)
C5	0.0430 (14)	0.0333 (12)	0.0319 (11)	−0.0059 (11)	0.0020 (10)	−0.0070 (9)
C6	0.0336 (12)	0.0388 (13)	0.0311 (11)	−0.0018 (10)	0.0075 (10)	0.0042 (9)
C7	0.0262 (11)	0.0297 (11)	0.0253 (10)	−0.0007 (9)	0.0032 (8)	0.0046 (8)
C8	0.0238 (10)	0.0311 (11)	0.0286 (10)	−0.0031 (9)	0.0015 (8)	0.0048 (8)
C9	0.0264 (11)	0.0318 (11)	0.0290 (10)	−0.0064 (9)	−0.0001 (9)	0.0012 (8)
C10	0.0286 (11)	0.0305 (11)	0.0309 (11)	−0.0021 (9)	−0.0001 (9)	0.0012 (9)
C11	0.0268 (11)	0.0338 (12)	0.0355 (11)	−0.0005 (9)	−0.0025 (9)	0.0001 (9)
C12	0.0291 (11)	0.0341 (12)	0.0279 (10)	−0.0030 (9)	−0.0011 (9)	0.0000 (9)
C13	0.0300 (12)	0.0293 (12)	0.0472 (13)	−0.0007 (10)	−0.0031 (10)	−0.0046 (10)
C14	0.0399 (16)	0.0565 (19)	0.077 (2)	0.0120 (14)	0.0149 (15)	0.0303 (16)
C15	0.0522 (18)	0.063 (2)	0.0633 (18)	0.0242 (15)	0.0275 (15)	0.0178 (15)
C16	0.0483 (18)	0.073 (2)	0.074 (2)	0.0206 (16)	0.0344 (16)	0.0345 (18)

Geometric parameters (Å, °)

S1—O2	1.4316 (18)	C6—H6A	0.9800
S1—O1	1.4351 (17)	C6—H6B	0.9800
S1—N3	1.644 (2)	C6—H6C	0.9800
S1—C7	1.748 (2)	C7—C12	1.392 (3)
O3—C14	1.439 (3)	C7—C8	1.396 (3)
O3—C13	1.442 (3)	C8—C9	1.379 (3)
N1—C1	1.334 (3)	C8—H8	0.9500

electronic reprint

N1—C2	1.353 (3)	C9—C10	1.407 (3)
N2—C1	1.340 (3)	C9—H9	0.9500
N2—C4	1.346 (3)	C10—C11	1.405 (3)
N3—C1	1.394 (3)	C11—C12	1.377 (3)
N3—H3N	0.8800	C11—H11	0.9500
N4—C10	1.377 (3)	C12—H12	0.9500
N4—C13	1.420 (3)	C13—C16	1.494 (4)
N4—H4N	0.8800	C13—H13	1.0000
C2—C3	1.384 (3)	C14—C15	1.488 (4)
C2—C5	1.500 (3)	C14—H14A	0.9900
C3—C4	1.386 (3)	C14—H14B	0.9900
C3—H3	0.9500	C15—C16	1.523 (4)
C4—C6	1.490 (3)	C15—H15A	0.9900
C5—H5A	0.9800	C15—H15B	0.9900
C5—H5B	0.9800	C16—H16A	0.9900
C5—H5C	0.9800	C16—H16B	0.9900

O2—S1—O1	119.23 (10)	C8—C7—S1	121.15 (17)
O2—S1—N3	109.23 (10)	C9—C8—C7	120.0 (2)
O1—S1—N3	102.72 (10)	C9—C8—H8	120.0
O2—S1—C7	108.79 (11)	C7—C8—H8	120.0
O1—S1—C7	108.43 (10)	C8—C9—C10	120.7 (2)
N3—S1—C7	107.85 (10)	C8—C9—H9	119.6
C14—O3—C13	107.26 (19)	C10—C9—H9	119.6
C1—N1—C2	115.27 (19)	N4—C10—C11	122.3 (2)
C1—N2—C4	115.51 (18)	N4—C10—C9	118.9 (2)
C1—N3—S1	125.67 (16)	C11—C10—C9	118.6 (2)
C1—N3—H3N	116.9	C12—C11—C10	120.4 (2)
S1—N3—H3N	115.6	C12—C11—H11	119.8
C10—N4—C13	124.3 (2)	C10—C11—H11	119.8
C10—N4—H4N	119.7	C11—C12—C7	120.4 (2)
C13—N4—H4N	112.9	C11—C12—H12	119.8
N1—C1—N2	128.24 (19)	C7—C12—H12	119.8
N1—C1—N3	117.29 (19)	N4—C13—O3	108.7 (2)
N2—C1—N3	114.47 (18)	N4—C13—C16	116.1 (2)
N1—C2—C3	121.2 (2)	O3—C13—C16	103.8 (2)
N1—C2—C5	116.0 (2)	N4—C13—H13	109.3
C3—C2—C5	122.79 (19)	O3—C13—H13	109.3
C2—C3—C4	118.66 (19)	C16—C13—H13	109.3
C2—C3—H3	120.7	O3—C14—C15	108.2 (2)
C4—C3—H3	120.7	O3—C14—H14A	110.1
N2—C4—C3	121.1 (2)	C15—C14—H14A	110.1
N2—C4—C6	116.5 (2)	O3—C14—H14B	110.1
C3—C4—C6	122.36 (19)	C15—C14—H14B	110.1
C2—C5—H5A	109.5	H14A—C14—H14B	108.4
C2—C5—H5B	109.5	C14—C15—C16	101.9 (2)
H5A—C5—H5B	109.5	C14—C15—H15A	111.4
C2—C5—H5C	109.5	C16—C15—H15A	111.4
H5A—C5—H5C	109.5	C14—C15—H15B	111.4
H5B—C5—H5C	109.5	C16—C15—H15B	111.4

electronic reprint

C4—C6—H6A	109.5	H15A—C15—H15B	109.3
C4—C6—H6B	109.5	C13—C16—C15	101.4 (2)
H6A—C6—H6B	109.5	C13—C16—H16A	111.5
C4—C6—H6C	109.5	C15—C16—H16A	111.5
H6A—C6—H6C	109.5	C13—C16—H16B	111.5
H6B—C6—H6C	109.5	C15—C16—H16B	111.5
C12—C7—C8	119.9 (2)	H16A—C16—H16B	109.3
C12—C7—S1	118.99 (17)		

O2—S1—N3—C1	56.4 (2)	N3—S1—C7—C8	95.14 (19)
O1—S1—N3—C1	−176.07 (18)	C12—C7—C8—C9	0.4 (3)
C7—S1—N3—C1	−61.7 (2)	S1—C7—C8—C9	−179.57 (16)
C2—N1—C1—N2	0.0 (3)	C7—C8—C9—C10	−0.1 (3)
C2—N1—C1—N3	179.72 (19)	C13—N4—C10—C11	−22.4 (4)
C4—N2—C1—N1	0.5 (3)	C13—N4—C10—C9	161.0 (2)
C4—N2—C1—N3	−179.25 (19)	C8—C9—C10—N4	176.2 (2)
S1—N3—C1—N1	1.0 (3)	C8—C9—C10—C11	−0.6 (3)
S1—N3—C1—N2	−179.18 (16)	N4—C10—C11—C12	−175.7 (2)
C1—N1—C2—C3	−0.2 (3)	C9—C10—C11—C12	0.9 (4)
C1—N1—C2—C5	179.3 (2)	C10—C11—C12—C7	−0.6 (4)
N1—C2—C3—C4	−0.1 (3)	C8—C7—C12—C11	0.0 (3)
C5—C2—C3—C4	−179.6 (2)	S1—C7—C12—C11	179.91 (18)
C1—N2—C4—C3	−0.8 (3)	C10—N4—C13—O3	−104.0 (3)
C1—N2—C4—C6	−179.43 (19)	C10—N4—C13—C16	139.5 (3)
C2—C3—C4—N2	0.6 (3)	C14—O3—C13—N4	−153.7 (2)
C2—C3—C4—C6	179.2 (2)	C14—O3—C13—C16	−29.5 (3)
O2—S1—C7—C12	156.85 (17)	C13—O3—C14—C15	5.4 (3)
O1—S1—C7—C12	25.8 (2)	O3—C14—C15—C16	20.2 (4)
N3—S1—C7—C12	−84.79 (19)	N4—C13—C16—C15	160.5 (3)
O2—S1—C7—C8	−23.2 (2)	O3—C13—C16—C15	41.3 (3)
O1—S1—C7—C8	−154.31 (18)	C14—C15—C16—C13	−37.0 (4)

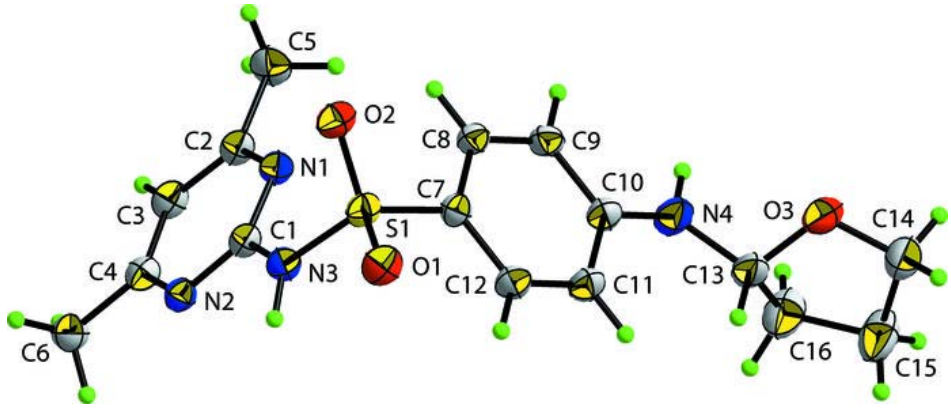
Hydrogen-bond geometry (Å, °)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
N3—H3n···O3^{i}	0.88	1.98	2.854 (3)	174
N4—H4n···N2^{ii}	0.88	2.22	3.086 (3)	167

Symmetry codes: (i) *x*, *y*−1, *z*; (ii) *x*, *y*+1, *z*.

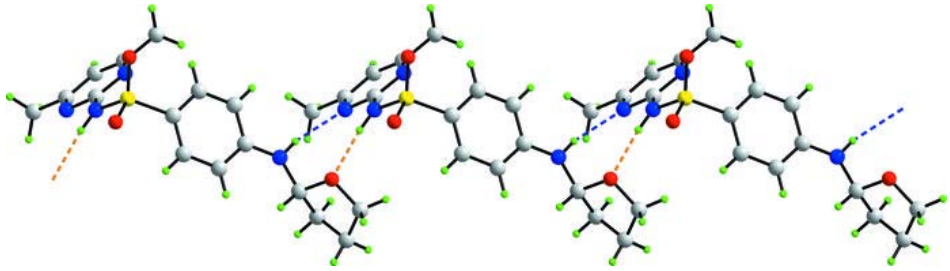
electronic reprint

Fig. 1



electronic reprint

Fig. 2



electronic reprint